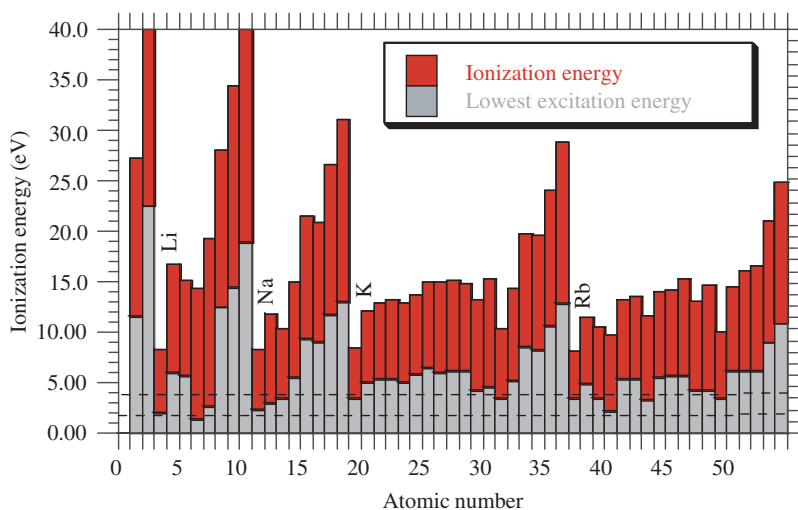
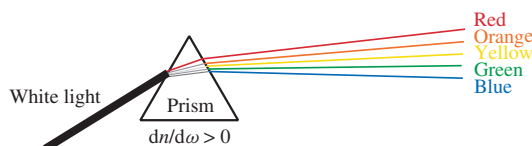


radio	microwave	IR	Visible	UV	X-ray	$\gamma$ -ray
$1-10^9$	$10^9-10^{12}$	$10^{12}-4 \times 10^{14}$	$4 \times 10^{14}-8 \times 10^{14}$	$8 \times 10^{14}-10^{17}$	$10^{17}-10^{20}$	$10^{20}$ up

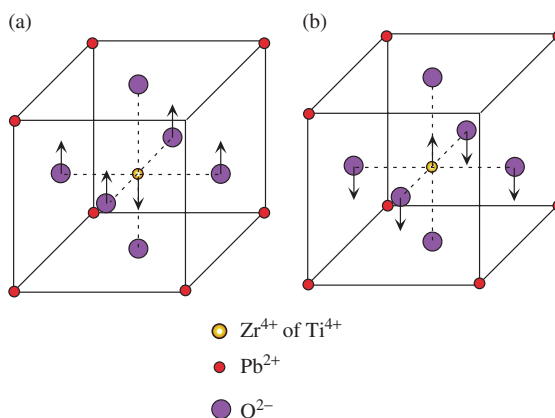
**Figure 1.1** The electromagnetic spectrum divided into regimes and frequencies within these regimes. Frequency is in units of Hz (i.e., cycles/s).



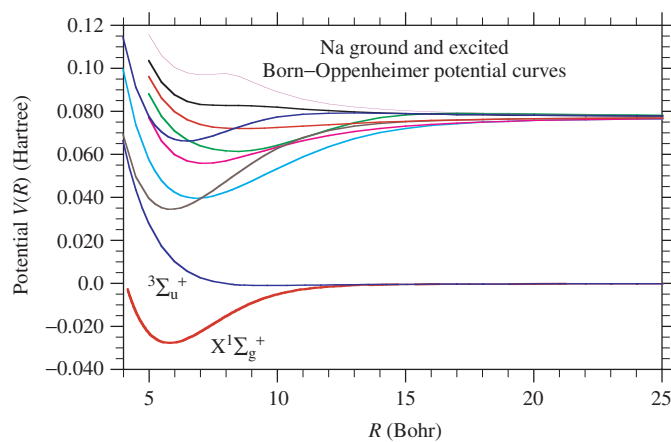
**Figure 2.3** Lowest electronic excitation energy and the ionization energy of the first 50 atoms in the periodic table. The visible region of the spectrum is indicated by the dashed lines. [See replacement file in "errors".]



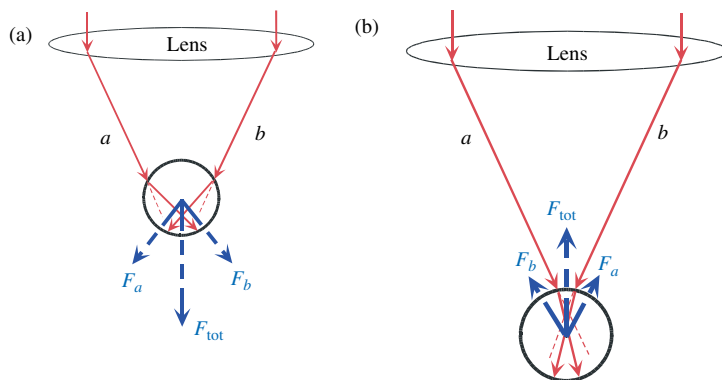
**Figure 2.8** Dispersion of white light by a glass prism.



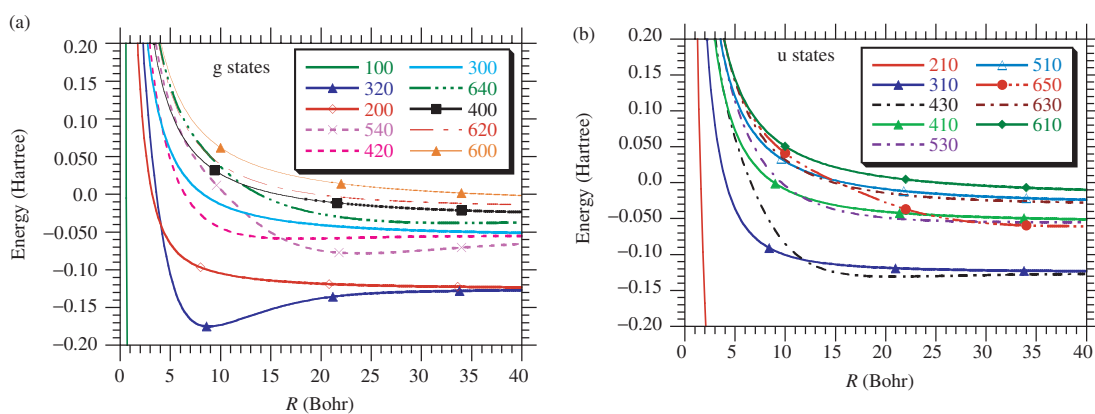
**Figure 2.34** Unit cell of the cubic perovskite  $Pb(Zr_xTi_{1-x})O_3$  crystal, with polarization (a) up or (b) down.



**Figure 3.6** All the adiabatic Born–Oppenheimer ground and excited molecular potential energy curves of sodium.



**Figure 3.14** Refraction of a light beam giving rise to optical tweezers. (a) Focus below the object center. (b) Focus above the object center.



**Figure 6.8**  $\text{H}_2^+$  excited potential curves for as a function of internuclear distance  $R$ . Gerade (g) states and ungerade (u) states. The states are labelled by their united atom quantum numbers  $n/l$  (see Section 6.3.5).

H																	He				
Li	sc Be															B	C	N	O	F	Ne
Na	Mg															sc Al	sc Si	P para or dia	S	Cl	Ar
K	Ca	Sc	sc Ti	sc V	Cr anti	Mn anti	Fe ferro	Co ferro	Ni ferro	Cu	sc Zn	sc Ga	Ge	As dia	Se dia	Br dia	Kr dia				
Rb	Sr	Y	sc Zr	sc Nb	sc Mo	Tc	sc Ru	Rh	Pd	Ag	sc Cd	sc In	sc Sn	Sb para	Te dia	I dia	Xe dia				
Cs	Ba	*	sc Hf	sc Ta	sc W	sc Re	sc Os	sc Ir	Pt	Au	sc Hg	sc Tl	sc Pb	sc Bi	Po	At	Rn dia				
Fr	Ra	**	Rt																		

sc
El
para

← Superconducting

← Symbol

← Magnetic property

*	La	Ce para or dia	Pr para	Nd anti	Pm	Sm anti	Eu ferro	Gd ferro	Tb ferro	Dy ferro	Ho ferro	Er ferro	Tm ferri	Yb	Lu para
**	Ac	sc Th para	Pa	sc U para	Nd	Pu anti	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lw

Figure 4.2 Magnetic properties of the elements at low temperatures.

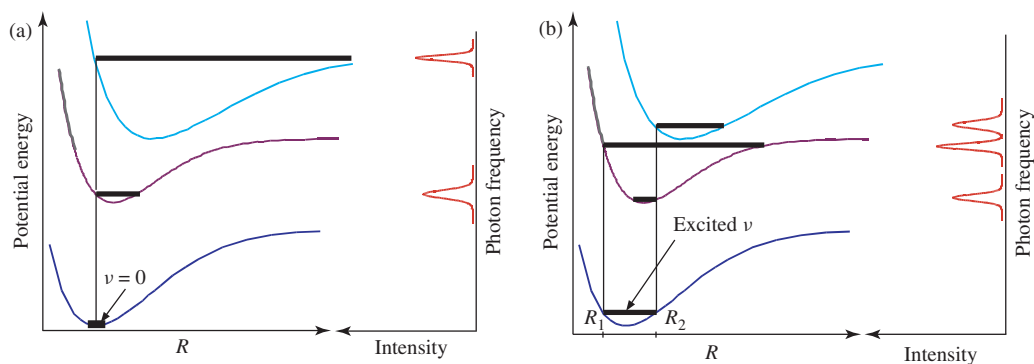
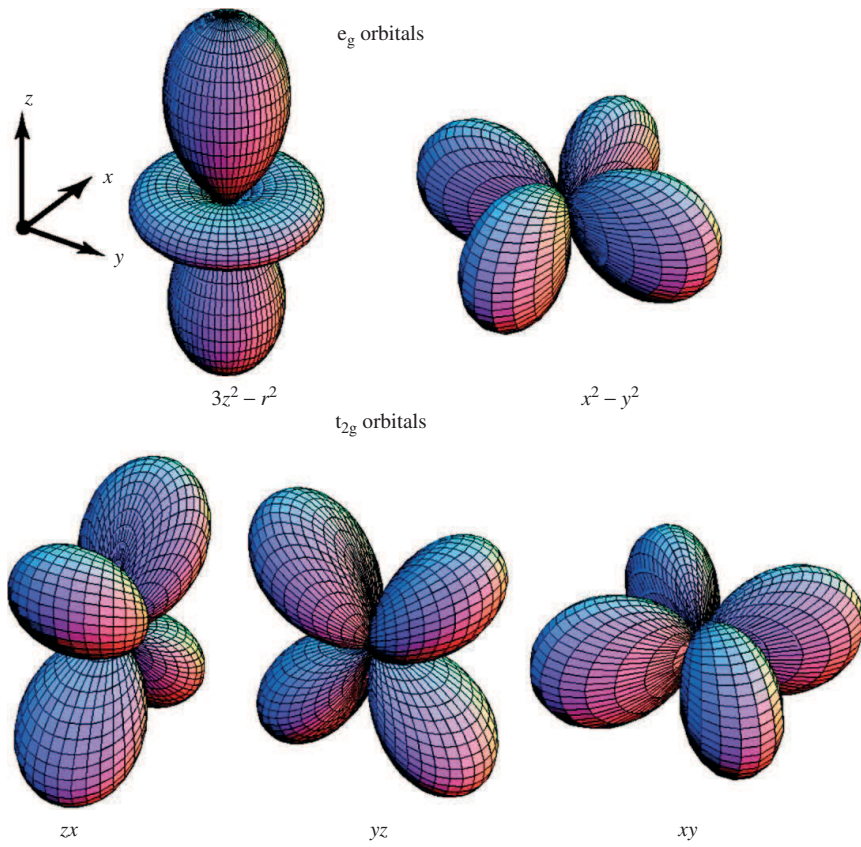
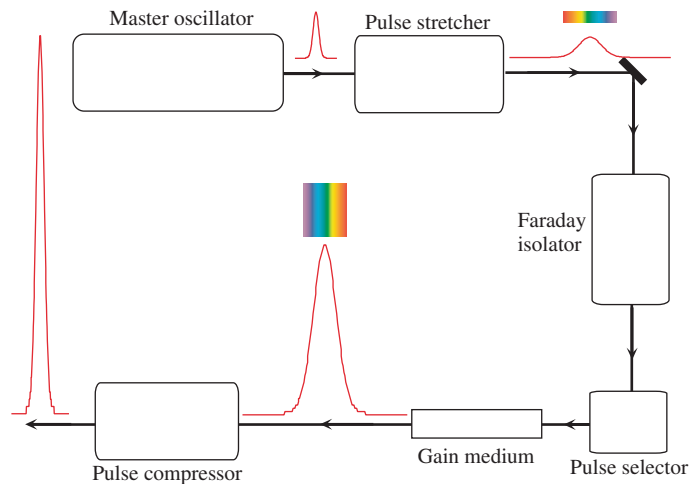


Figure 6.12 Schematic illustration of the Franck–Condon principle. (a) Absorption from the ground vibrational state of the ground electronic potential. (b) Absorption from an excited vibrational state of the ground electronic potential. The points labelled  $R_1$  and  $R_2$  are the turning points of the excited vibrational state on the ground electronic potential.



**Figure 6.22** The five d orbital functions.



**Figure 7.10** Chirped pulse amplifier system.

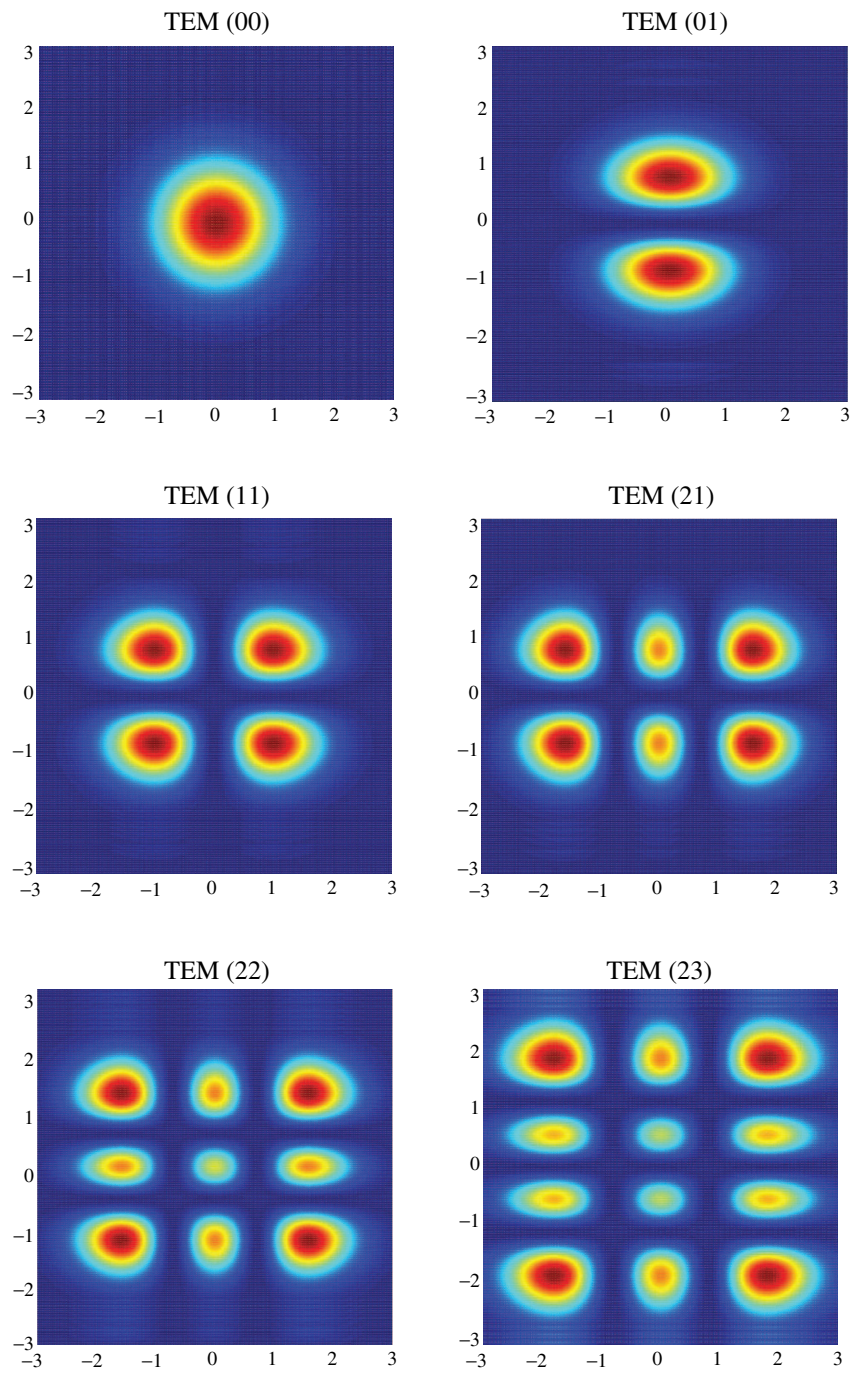
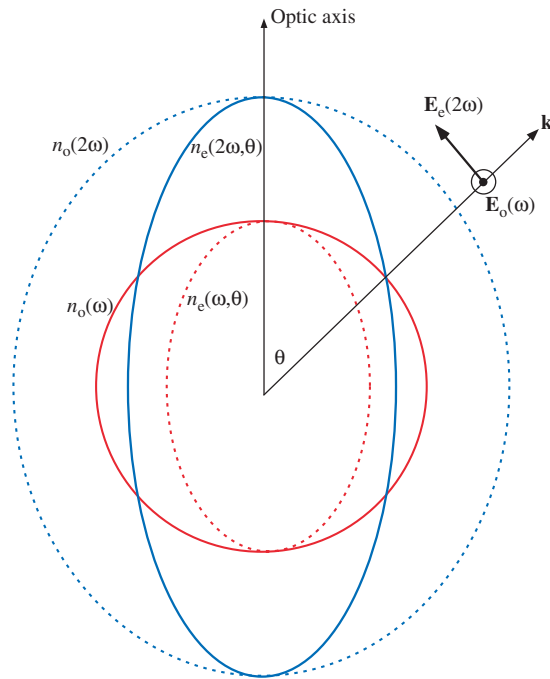
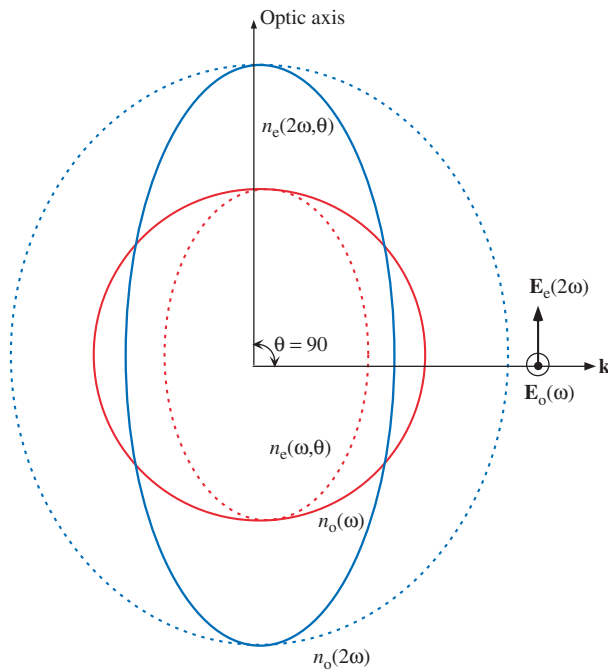


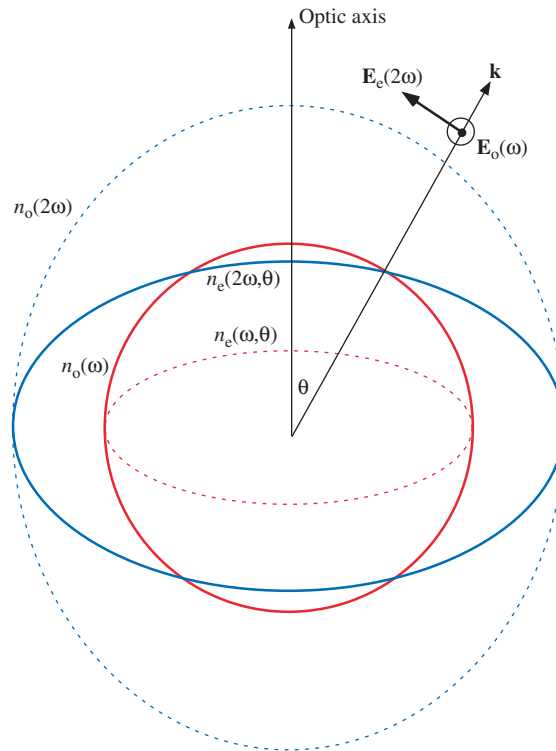
Figure 7.12 Transverse Hermite-gaussian spatial intensity profiles.



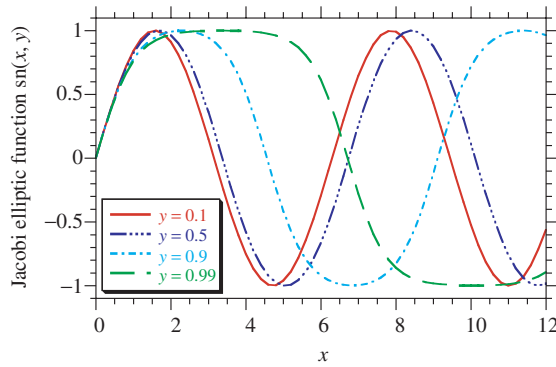
**Figure 8.1** Indices of refraction for the fundamental and the second harmonic in a negative uniaxial crystal ( $n_o > n_e$ ) with type I phase-matching.



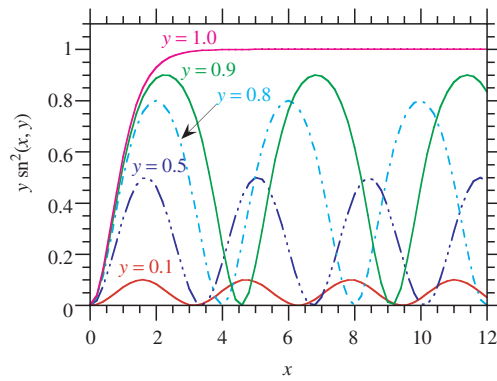
**Figure 8.2** Type I phase-matching with  $90^\circ$  matching.



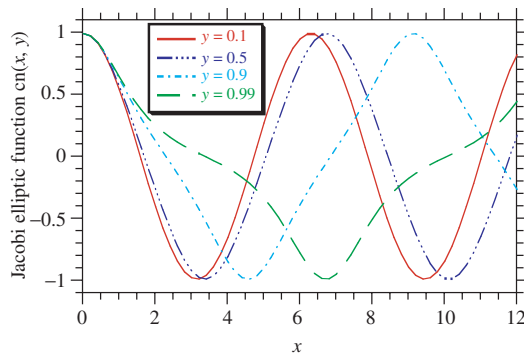
**Figure 8.3** Indices of refraction for the fundamental and the second harmonic in a positive uniaxial crystal ( $n_e > n_o$ ) with type I phase-matching.



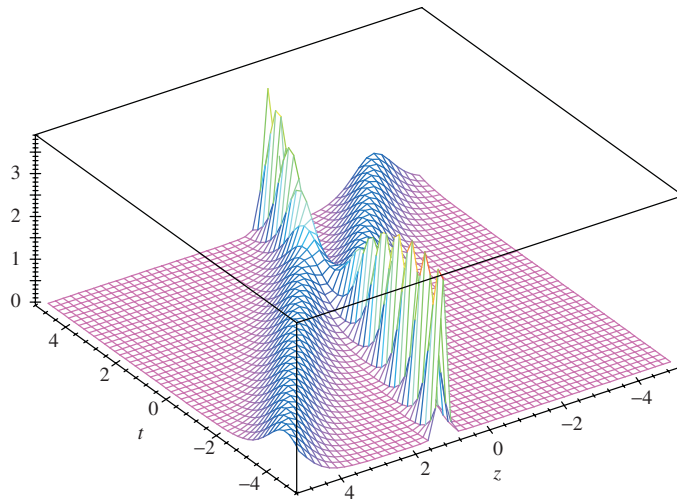
**Figure 8.8** Jacobi elliptic function  $\text{sn}(x, y)$  vs  $x$  for several values of  $y$ .



**Figure 8.9** The product  $y \operatorname{sn}^2(x, y)$  vs  $x$  for several values of  $y$  models the output of the SFG vs propagation distance in the crystal for various ratios of input field strengths.



**Figure 8.14** Jacobi elliptic function  $\operatorname{cn}(x, y)$  vs  $x$  for several values of  $y$ .



**Figure 8.20** (a) Two-soliton solution for the higher-order NSE, Eq. (8.100). (b) Magnitude of the black soliton  $u(z, t)$  in Eq. (8.112) with  $B = 1$ , and  $A = 1.0, 1.5$  and  $2.0$ .